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# Ammonium chloride Cell Culture Tested

Product Number **A0171** Store at Room Temperature

### **Product Description**

Molecular Formula: NH<sub>4</sub>Cl Molecular Weight: 53.49 CAS Number: 12125-02-9

Synonyms: ammonium muriate, sal ammoniac,

salmiac1

This product is cell culture tested and is appropriate for use in cell culture experiments.

Ammonium chloride is a reagent that is used in a variety of industrial and research applications. Industrial uses include electroplating, tinning, and the manufacture of dyes. It is also used as a fluxing agent for the galvanizing of steel, the refinement of zinc, and the coating of sheet iron with zinc.<sup>1</sup>

In biological research, ammonium chloride is often used for the lysis of human red blood cells. <sup>2,3,4</sup> Ammonium chloride has been used in the study of basic calcium phosphate crystals in fibroblasts. <sup>5</sup> The use of ammonium chloride in the isolation of proteins from 50S ribosomal subunits of *Bacillus stearothermophilus* has been described. <sup>6</sup> A study of the nucleic acid binding protein HSP15 that uses ammonium chloride to investigate the effect of different salt conditions on HSP15 binding to 50S subunits has been published. <sup>7</sup>

A differential pulse voltammetry procedure for the detection of copper, lead, cadmium, and nickel in environmental matrices that uses ammonia-ammonium chloride buffer has been reported. 8

### **Precautions and Disclaimer**

For Laboratory Use Only. Not for drug, household or other uses.

## **Preparation Instructions**

This product is soluble in water (100 mg/ml. It is also soluble in methanol and ethanol. The pH of various percentage solutions of NH<sub>4</sub>Cl has been reported:

1% solution = pH 5.5 3% solution = pH 5.1 10% solution = pH  $5.0^1$ 

#### References

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